

Utilization of 1-(3-aminopropyl)-imidazole in development of pH-sensitive nanocarrier for anticancer drug delivery

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Abstract

pH-sensitive nanocarriers have demonstrated as successful drug carriers due to the higher drug accumulation at the tumor site by pH-dependent release at acidic tumor microenvironment. Imidazole can be utilized to obtain pH-sensitive nanocarriers. One of the imidazole-based compounds used by researchers in development of pH-responsive carriers is 1-(3-Aminopropyl)-imidazole. This review is focused on utilization of 1-(3-aminopropyl)-imidazole to develop pH-sensitive nanocarrier especially for delivery of anticancer drug. In addition, this review also shows research performed by researchers to develop pH-sensitive nanocarrier for anticancer drug delivery using 1-(3-aminopropyl)-imidazole to prepare imidazole-based pH-sensitive nanocarriers. The data were collected from published journals with 15 and 20 journals as primary and supporting literatures, respectively. Properties of imidazole groups to deprotonate at pH 7.4 and protonate at pH \leq 6.5 implies that the drug release was pH-dependent, leading to a limited release of drug from carriers under physiological pH conditions and diminishing the drug's effect in blood circulation before reaching the tumor site, resulting in more effective anticancer activity.

Keywords: pH-sensitive, imidazole, poly (β -Benzyl-L-Aspartate), polymer, nanocarrier

1. Introduction

Delivery systems of drug using pH-sensitive carrier have shown improved anticancer activity in comparison with pH-insensitive carrier (1, 2). pH-sensitive nanocarriers have shown great interest because of their advanced functionality (3-6). Different characteristics were shown by the pH-responsive nanocarriers under an acidic, basic, or neutral environment (7, 8). In the case of anticancer drug delivery, pH-sensitive carriers have been reported as

competent drug carrier as pH-dependent release of anticancer agents triggered by tumor microenvironments (tumor extracellular pH (pHex = \sim 6.5 to 7.2) or endosomal pH (pHen \leq 6.5) (3-6).

Imidazole is a typical pH-sensitive compound, which was used to obtain the pH-sensitive and sustained drug release ability of nanoparticles (9). 1-(3-aminopropyl)-imidazole has been used in development of pH-sensitive nanocarrier. Imidazole can protonate resulted in a

positive charge under acidic pH condition (10). Deprotonation at physiological pH (pH 7.4) and protonation at $\text{pH} \leq 6.5$ of imidazole groups provided pH triggered drug release and tumor pH targeted drug delivery. Some methods can be used to synthesize pH-sensitive nanocarrier using 1-(3-aminopropyl)-imidazole either by grafting or by directly substitution/modulation of functional side groups of a polymer (11, 12).

This review is aimed to depict utilization of 1-(3-aminopropyl)-imidazole in development of pH-sensitive nanocarrier including properties of imidazole group as pH-sensitive moiety and synthesis methods to prepare pH-sensitive nanocarrier using 1-(3-aminopropyl)-imidazole. In addition, this review also shows several studies performed by researchers that use 1-(3-

aminopropyl)-imidazole in development of pH-responsive nanocarrier to deliver anticancer drug. Urgency of this review paper is to introduce the use 1-(3-aminopropyl)-imidazole in preparation of pH-sensitivity carrier using easy and simple method.

2. Methodology

By using specific keywords “imidazole, 1-(3-aminopropyl)-imidazole pH-sensitive nanocarrier for cancer”, this review was prepared. Inclusions criteria (related to specific keywords) and exclusions criteria (opinions and unrelated topics) were also determined. Finally, 15 and 20 journals published in 2011-2022 were collected as primary and supporting literatures, respectively. Figure 1 depicts the flowchart of the methodology.

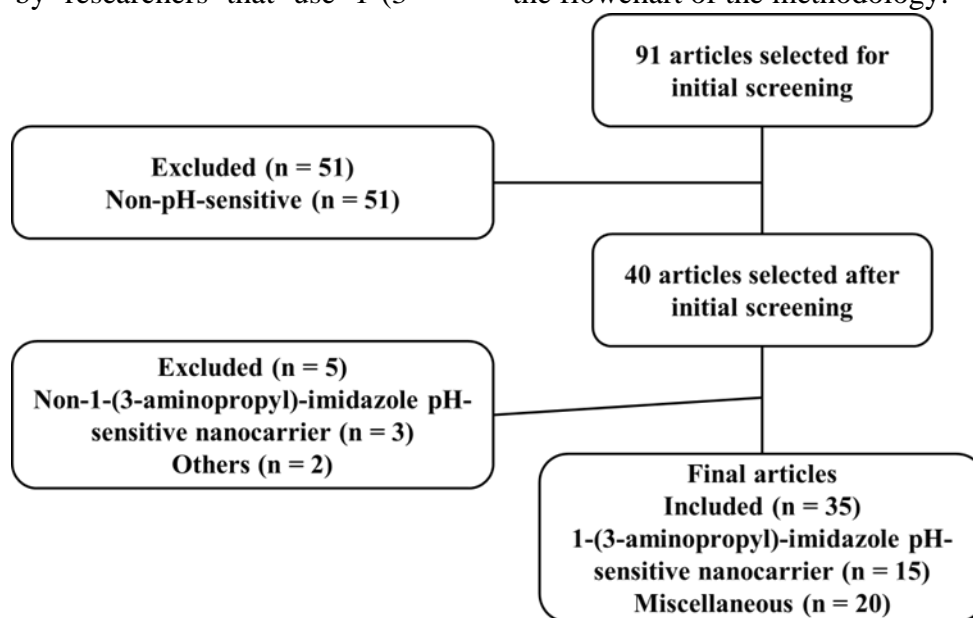


Figure 1. Flowchart of methodology

2.1 Properties of imidazole group as pH-sensitive moiety

Acidic microenvironment is abnormality of tumor tissue (10). Intracellular pH (pH_i) of normal cells in healthy tissues is 7.2 and a slightly higher extracellular pH (pH_{ex}) of 7.4. However, a pH_i of 7.2 and pH_{en} of 6.2–7.0 are found in tumor cells (13-16). The lower pH_{en} in the tumor interstitium is caused by lactate accumulation. Elevated anaerobic

metabolism by the cancer cells produces an acidic by-product called lactate in the hypoxic tumor microenvironment (17-19). Therefore, pH in tumor microenvironment or inside cancer cells is one of the internal stimuli for cancer targeting (7, 14). pH-responsive nanocarriers containing weakly acidic or basic ionizable groups (pK_a values 3 - 10) can either donate or accept protons in response to changes in environmental pH, resulting in change in

structural and other properties (surface activity, solubility, chain conformation, etc.) (7, 8, 20). Thus, pH activation of

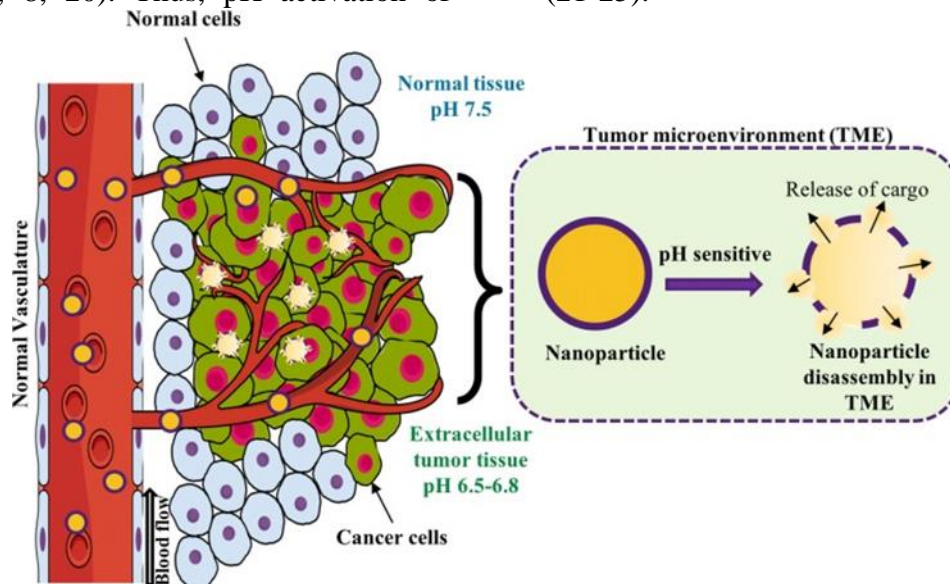


Figure 2. Schematic illustration of pH activation of nanoparticle by tumor microenvironment (21)

One of the imidazole-based compounds used by researchers in development of pH-responsive carriers is 1-(3-Aminopropyl)imidazole (Figure 3) which is used in the synthesis of pH-responsive polyaspartamide derivatives and the preparation of pH-sensitive amphiphilic polymers (26). Imidazole exhibits a prominent pH-sensitive functional group in pharmaceutical sciences. pKa value of imidazole is 5.0–6.5, leading to protonate under acidic pH, resulting in a positive charge (Figure 4) (10-12, 27).

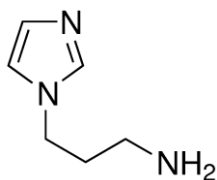


Figure 3. Chemical structure of 1-(3-Aminopropyl)imidazole (26)

2.2 Synthesis methods to prepare pH-sensitive nanocarrier using 1-(3-aminopropyl)-imidazole

Either by grafting or direct substitution/modulation of functional side

nanoparticle by tumor microenvironment can be used to release of drug (Figure 2) (21-25).

If the imidazole groups are grafted to a polymer or substituted /modulated of functional side groups of a polymer for a drug carrier, the disintegration of carrier is induced by pH change (11, 12). At pH 7.4, the imidazole groups are deprotonated. However, at $\text{pH} \leq 6.5$, imidazole groups protonate, leading to change in structure of carrier, resulting in release drug from its destabilized cores of the carrier (Figure 4) (11, 12, 28).

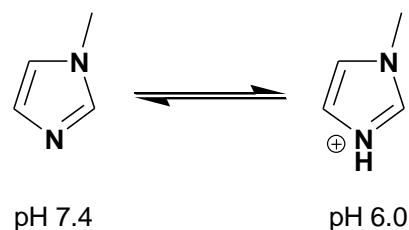


Figure 4. Illustration of protonation of imidazole groups under different pH conditions (pH 7.4 and 6.0)

groups method can be used for synthesis pH-sensitive nanocarrier using 1-(3-aminopropyl)-imidazole (11, 12, 29-31). For the grafting method, Kim *et al.* develop a pH-sensitive polymer by grafting the

imidazole groups to poly(aspartic acid)-*block*-poly(ethylene glycol) (P(Asp)-PEG), resulting in P(Asp-*g*-Im)-PEG to target acidic pH (Figure 5) Briefly, poly(ethylene glycol)-poly(β -benzyl-L-aspartate) (PEG-PBLA) was prepared by coupling PBLA with monocarboxylated PEG. Next, the benzyl groups were removed from PBLA by dissolving the PBLA-PEG in a solution of dimethylformamide/methanol (DMF/MeOH) mix and Pd/C catalyst. To graft the imidazole groups, the preactivated of deprotected P(Asp)-PEG was dissolved

in DMF in the presence of 1-(3-aminopropyl)-imidazole and stirred for 24 h at 30 °C (11). Other researchers, Cheng *et al.* also developed pH-sensitive carboxymethyl chitosan nanoparticles by grafting *N*-(3-Aminopropyl)-imidazole onto carboxymethyl chitosan and showed pH-triggered drug release at acidic environment (29). In addition, Han *et al.* synthesized a pH-responsive nanocarrier using hyaluronic acid (HA)-*graft*-imidazole-dodecylamine (HID) for anticancer drugs delivery (32).

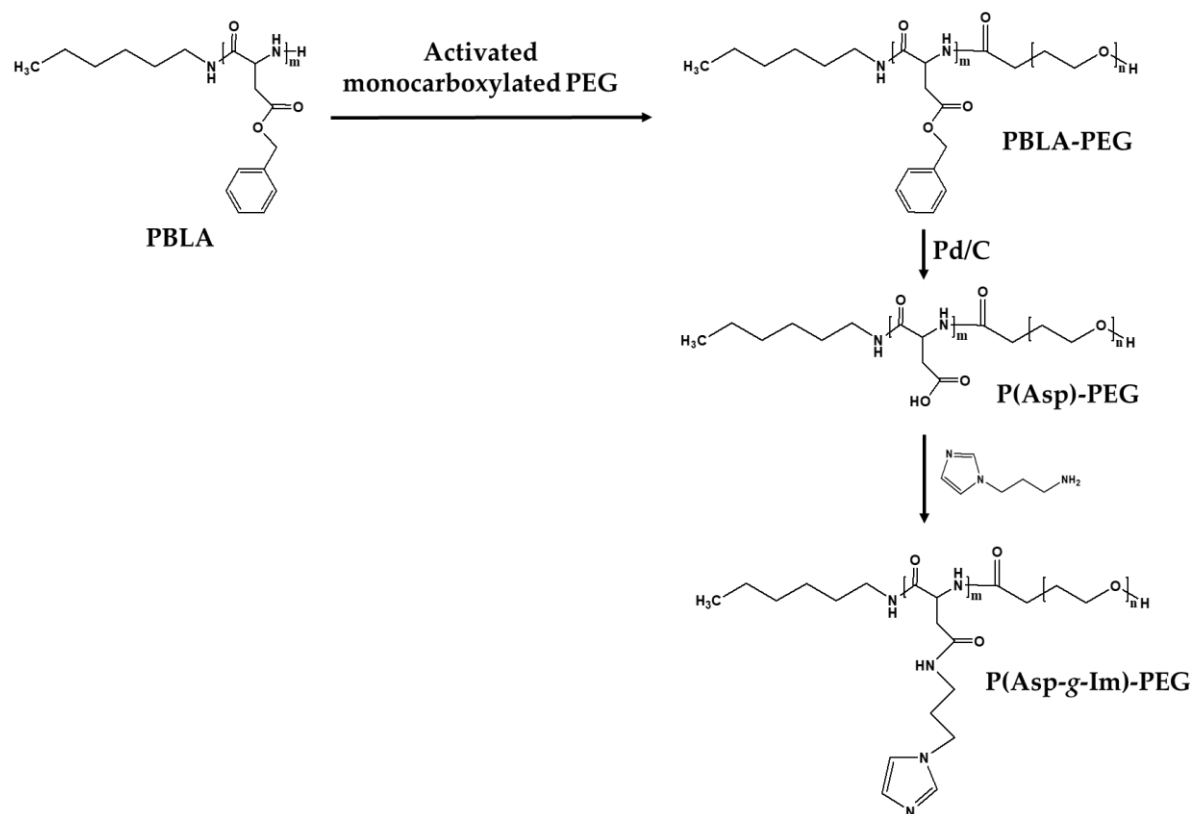


Figure 5. Synthetic route of grafting the imidazole group to P(Asp)-PEG (11)

For substitution/modulation of functional side groups method, Chu *et al.* prepared pH and reduction micelles using methoxy-poly(ethylene glycol)-*b*-poly[(benzyl-L-aspartate)-*co*-(*N*-(3-aminopropyl)imidazole-L-aspartamide)] (mPEG-SS-P(BLA-*co*-APILA). These researchers introduced imidazole groups in the side

chain of the block copolymer for pH sensitivity moiety by substitution reaction from mPEG-SS-PBLA, resulting in mPEG-SS-P(BLA-*co*-APILA (Figure 6). The block copolymer and 1-(3-aminopropyl)-imidazole was mixed in DMF under stirring at 40 °C for 5 h (33).

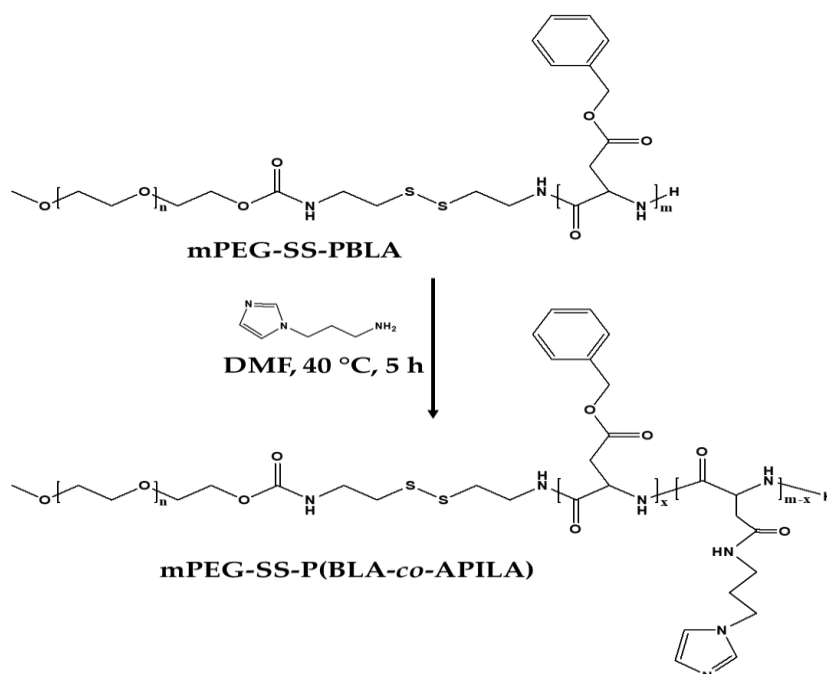


Figure 6. Substitution reaction of mPEG-SS-PBLA to prepare mPEG-SS-P(BLA-co-APILA) (33)

Recent study by Sim et al. also developed a pH-sensitive polymer prepared by partly modulation of benzyl groups of the hydrophobic PBLA block of PBLA-PEG. Dissolving PBLA-PEG in DCM containing 1-(3-aminopropyl)-imidazole and

triethylamine (TEA) at 35 °C under stirring for 24 h would result in poly[(benzyl-L-aspartate)-co-(N-(3-aminopropyl)imidazole-L-aspartamide)]-poly(ethylene glycol) (PABI-PEG) (Figure 7) (12).

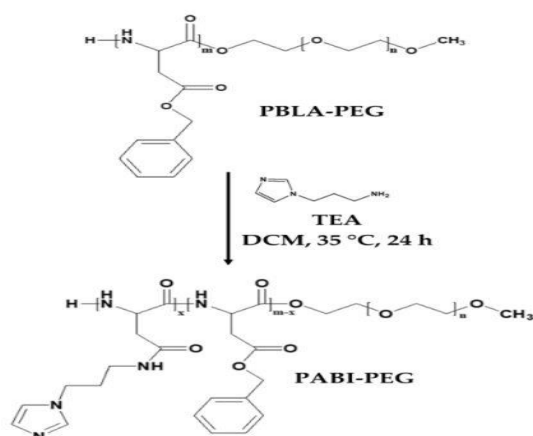


Figure 7. Modulation of functional side groups of PEG-PBLA to form PEG-PABI (12)

2.3 Utilization of 1-(3-aminopropyl)-imidazole in development of pH-sensitive nanocarrier for anticancer drug delivery

A number of studies have used 1-(3-aminopropyl)-imidazole to develop pH-

responsive nanocarrier for delivery of an anticancer drug (28-31, 33-35). For example, Chu et al. prepared dual sensitive (pH and reduction) micelles. Doxorubicin (DOX) was loaded into the micelles. DOX release from micelles showed retarded

release in pH 7.4 and fast release at acidic or reductive environment. These researchers reported DOX was found rapid internalization and fast release inside cells. The cytotoxicity study using HeLa cells showed that the DOX-encapsulated micelles could kill cancer cells (33). Cheng *et al.* developed nanoparticles of surface-

fluorinated and pH-sensitive carboxymethyl chitosan (CMCS). To prepare pH-responsive nanoparticles, *N*-(3-aminopropyl)-imidazole (API) was grafted to CMCS. At acidic condition, the swelling of nanoparticles containing imidazole was obtained, resulting in the accelerated release of DOX (29).

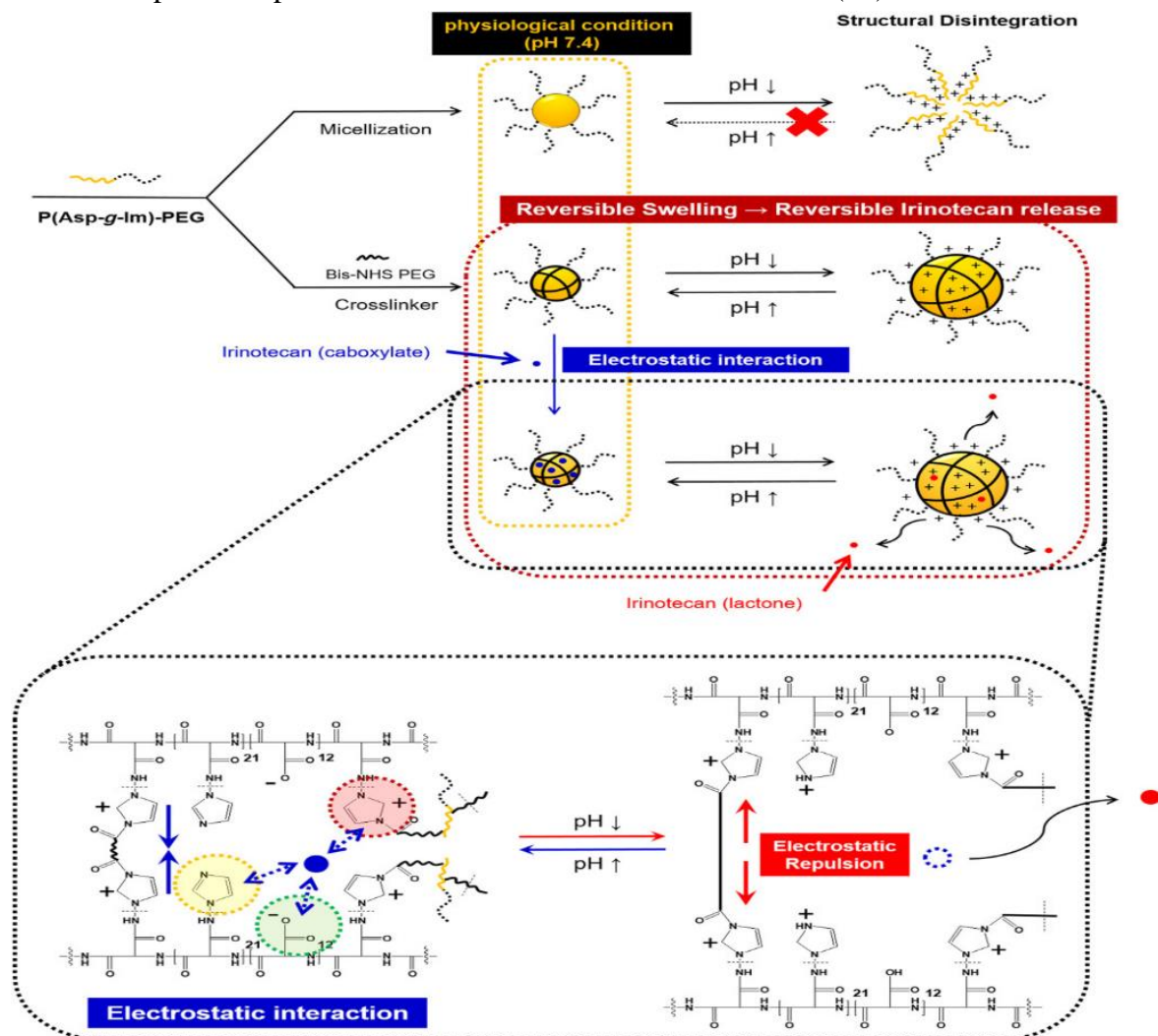


Figure 8. Schematic diagram of irinotecan-loaded pH-sensitive nanogels. Reproduced with permission of Sim T, Lim C, Cho YH, Lee ES, Youn YS, Oh KT., Development of pH-Sensitive Nanogels for Cancer Treatment using Crosslinked Poly(Aspartic Acid-*graft*-Imidazole)-*block*-Poly(Ethylene Glycol); published by John Wiley and Sons, 2018.

Oh's group developed pH-responsive nanogels using crosslinked P(Asp-g-Im)-PEG for cancer treatment (Figure 8). Irinotecan (IRI), an anticancer drug, was used as a model drug that was loaded into nanogels (ILNs). Study results indicated

that IRI release was higher under pH 6.5 than pH 7.4 condition. Further, the effect of ILNs on cell viability was also tested against colorectal cancer. Cytotoxic effect of the ILNs was higher at pH 6.5 than at pH 7.4 (34).

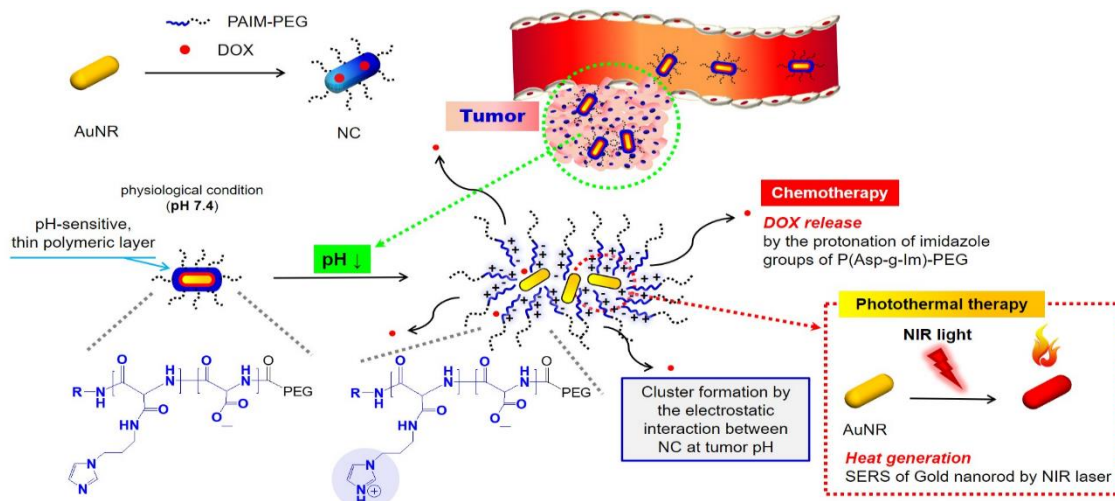


Figure 9. Illustration of pH-sensitive nanocluster (NC) system using a pH-sensitive polymer, P(Asp-g-Im)-PEG, for encapsulation of gold nanorods and DOX for cancer treatment (28).

In other studies, Oh’s group prepared pH-responsive nanocluster (NC) using a pH-responsive polymer. Gold nanorods and DOX were encapsulated by P(Asp-g-Im)-PEG for cancer treatment (Figure 9). These researchers reported not only a stable and structured but also low systemic toxicity at pH 7.4 of NC systems resulted. In contrast,

aggregated structures of the NC systems were occurred at pH 6.5, leading to high drug release. Furthermore, the NC systems also resulted an increased accumulation and high DOX release at the tumor site at pH_{ex} and pH_{en} with application of near-infrared light locally, leading to increase antitumor efficacy (Figure 10) (28).

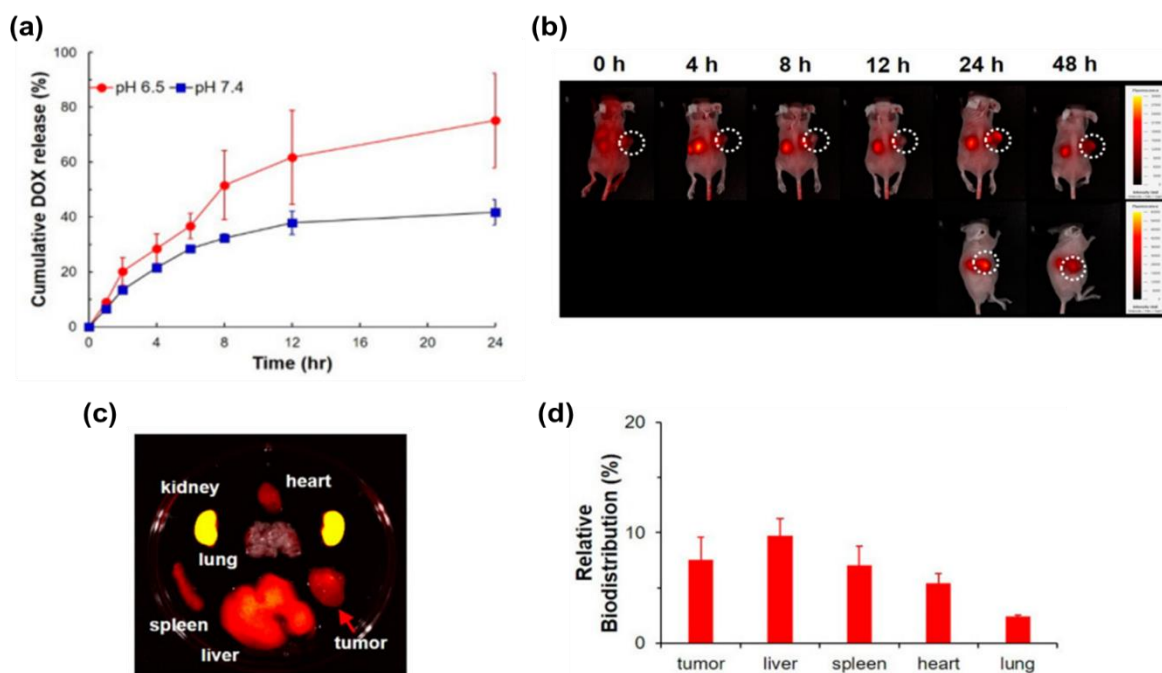


Figure 10. (a) DOX release at different pH conditions (pH 7.4 and pH 6.5). (b) whole body imaging after i.v. injection. (c) EX vivo optical and fluorescent imaging of tumor and organs obtained 24 h post-injection. (d) relative biodistribution by quantitative fluorescence intensity (FI) of tumors and main organs. Modified from (28).

3. Conclusion

1-(3-aminopropyl)-imidazole has been used in development of pH-sensitive nanocarrier. pH activation of nanocarrier by tumor microenvironment can be used for anticancer drug delivery. Deprotonation at physiological pH (pH 7.4) and protonation at $\text{pH} \leq 6.5$ of imidazole groups implies that the drug release was pH-dependent, leading to minimal drug effect in the bloodstream before the drug-loaded carrier reach the tumor site. The tumor pH_{ex} and pH_{en} would trigger release of drug from its carrier, resulting in effective antitumor activity.

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